

Determination of Aflatoxin M₁ by High-Pressure Liquid Chromatography Using Fluorescence Detection

A high-pressure liquid chromatography method developed for aflatoxins B₁, B₂, G₁, and G₂ in various foods was extended to the determination of a residue metabolite, M₁, in dairy products. Products tested were fluid milk, sour cream, cottage cheese, and buttermilk. The method was found to be rapid, accurate, sensitive, and selective. M₁ was determined at levels as low as 0.3 ppb and confirmed by formation of the trifluoroacetic acid reaction product. Recoveries were 93-125% at 1 ppb in three products.

Aflatoxins, produced by certain fungi, occur naturally in a wide variety of foods. The most often encountered of the group, aflatoxin B₁, is a recognized carcinogen and shows other toxic characteristics. When B₁ is ingested by dairy cows, a portion of it is metabolized to M₁ which may be found in the milk. The metabolite is about 75% as toxic to ducklings as the parent compound (Wogan, 1969). A recent incident involving the wide distribution of cottonseed containing high levels of aflatoxin B₁ caused contamination of milk and milk products with M₁. Determination of M₁ at levels of 0.5 ppb or below in fluid milk and higher levels in other dairy products was necessary. The official (AOAC, 1979) procedure employing two-dimensional thin-layer chromatography (TLC) with visual or densitometric measurement is often complicated by fluorescent extractives not well separated from the M₁ (Van Egmond et al., 1978). A high-pressure liquid chromatography (LC) method (Takahashi, 1977; Beebe, 1978) which has been found by this laboratory to be sensitive and precise for the measurement of aflatoxins B₁, B₂, G₁, and G₂ was extended to include M₁.

The method employs reverse-phase LC separation, followed by fluorescence detection of the aflatoxins and/or their trifluoroacetic acid (TFA) reaction products. Treatment with TFA causes the addition of water across the double bond of the terminal furan ring in B₁ and G₁ to form B_{2a} and G_{2a}, respectively. Figure 1 illustrates these structural changes. This increases the fluorescence of the molecules in the reverse-phase solvent system and decreases their retention times (*t_R*). Although the exact configuration of the M₁/TFA reaction product has not yet been established, we have noted changes in fluorescence and *t_R* similar to those of B₁ and G₁ and therefore have assumed similar structural changes. The reaction product, tentatively designated M_{2a}, is three-four times more fluorescent than M₁. Under the conditions described below, we have found the reaction reproducible and quantitative.

EXPERIMENTAL SECTION

Sample Preparation. Dairy products (fluid milk, sour cream, cottage cheese, and buttermilk) were blended with acetone-water (3:1, moisture content of product included in water figure), filtered, treated with lead acetate, followed by sodium sulfate, and refiltered. The filtrate was washed with hexane and extracted with chloroform. Duplicate aliquots of chloroform extract, each equivalent to 20 mL of milk, were placed in vials and evaporated (AOAC, 1975).

Trifluoroacetic Acid Reaction. Hexane (200 μL) was added to the residue in each vial. To one portion, 50 μL of TFA was added, and the vial was capped, mixed, and allowed to react for 15 min. Both portions were then diluted with 5 mL of water-acetonitrile (9:1). The aqueous layers were withdrawn and filtered through a syringe-

Table I. Correlation of Naturally Occurring M₁ in Milk and B₁ in Cottonseed^a

source	milk M ₁ , ppb	cottonseed B ₁ , ppb	(M ₁ /B ₁) × 10 ⁻⁴
A	4.3	5690	7.6
B	0.68	1090	6.2
C	8.8	7400	11.9
D	6.7	7510	8.9
E	6.4	7050	9.1
F	5.0	8840	5.7

^a Cottonseed homogeneity and feeding rations unknown.

mounted Millipore filter (type LS). Inject in LC on the same day as derivatized.

Standard Solutions. M₁ (10 ng, Eureka Labs, Sacramento, CA) and 5 ng of M₂ (Sigma Chemical Co., St. Louis, MO) were placed in each of two vials, evaporated to dryness, and treated as above (hexane omitted).

LC System. The liquid chromatograph used was Spectra-Physics, (Santa Clara, CA) Model 3500B with dual pumps, solvent programmer, and 500-μL loop injector. The LC column (3.2 mm i.d. × 25 cm) was packed with 10 μm Spherisorb-ODS. Mobil phases were (A) water and (B) acetonitrile-methanol (3:2). Solvents were filtered through 5 μm Millipore filter and degassed under vacuum. Operating conditions were: temperature, ambient; flow rate, 3.2 mL/min (constant); initial pressure, 2300 psi (automatically varied); gradient, linear from 10% B to 20% B in 20 min (program initiated upon injection). Detection was by Aminco Fluoro-Monitor (American Instrument Co., Silver Spring, MD), excitation at 365 nm and detection above 400 nm, using 10-mV recorder and chart speed of 10 cm/h. Sensitivity was adjusted to give half-scale deflection with 1 ng of M_{2a} equivalent.

RESULTS AND DISCUSSION

Chromatograms in Figure 2 show the separation of the six aflatoxins we have studied. Note that when the TFA reaction is allowed to proceed for 15 min, the M₁, B₁, and G₁ are totally converted to M_{2a}, B_{2a}, and G_{2a}. However, if the reaction is stopped by dilution immediately after the TFA is added, the B₁ and G₁ are converted but the M₁ conversion is not yet detectable. These standard chromatograms include M₂, which became available to us quite recently. With present cleanup methods, M₂ is obscured in the sample chromatogram by a fluorescent artifact.

Figure 3 shows successive injections of TFA-treated standard, nontreated standard, TFA-treated sample, and nontreated sample. Comparison of peak heights allows quantitation and the peak shift provides identification. The change of peak height and retention time between the M₁ and M_{2a} in the treated and nontreated portions is visible even if the product contains some background fluorescent materials. Most of the dairy products analyzed

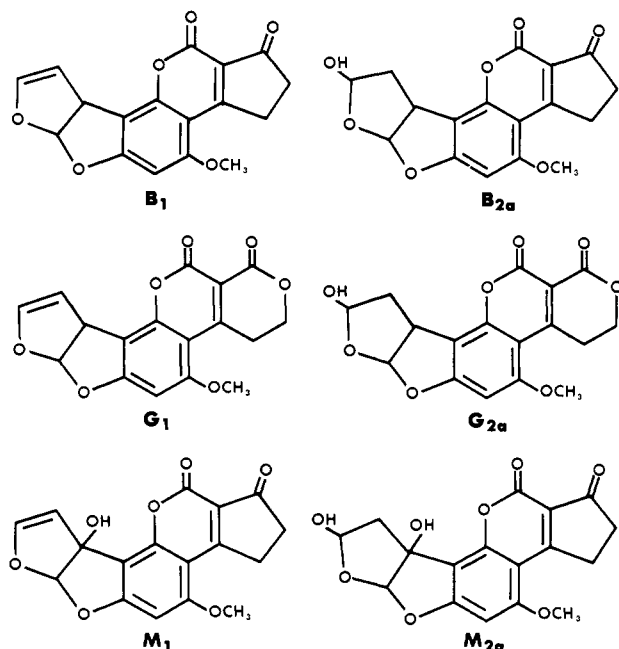


Figure 1. Chemical structures of some aflatoxins.

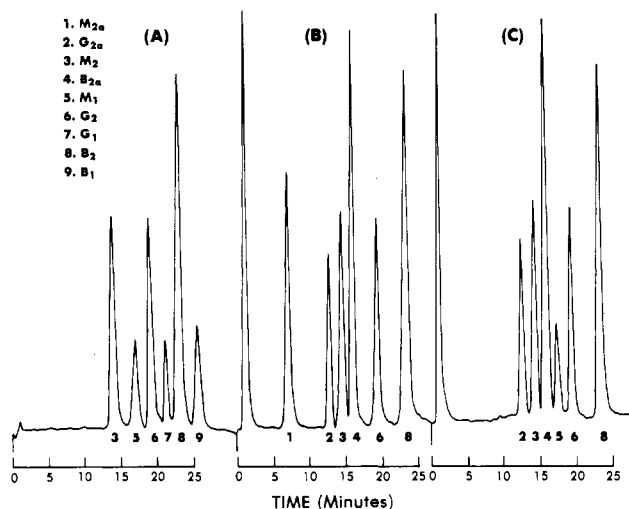


Figure 2. LC resolution of aflatoxin standards: (A) untreated with TFA, (B) 15-min TFA reaction, (C) 30-s TFA reaction. Amounts injected: (1, 3, 5) 1 ng, (2, 4, 6, 8) 0.5 ng, (7, 9) 15 ng each. LC parameters are as described in the text.

contained a fluorescent artifact which eluted just prior to the M₁. The height and t_R of this peak did not change with the addition of TFA. With isocratic elution, this artifact may interfere with M₁, requiring further cleanup of the sample extracts.

During the investigation of the aforementioned incident, a number of milk samples were analyzed for M₁ along with the associated cottonseed which had been fed at the rate of about 4 lb daily per cow along with other unknown feed for approximately one month. Table I lists the B₁ and M₁ contents of related samples. Although this was not a

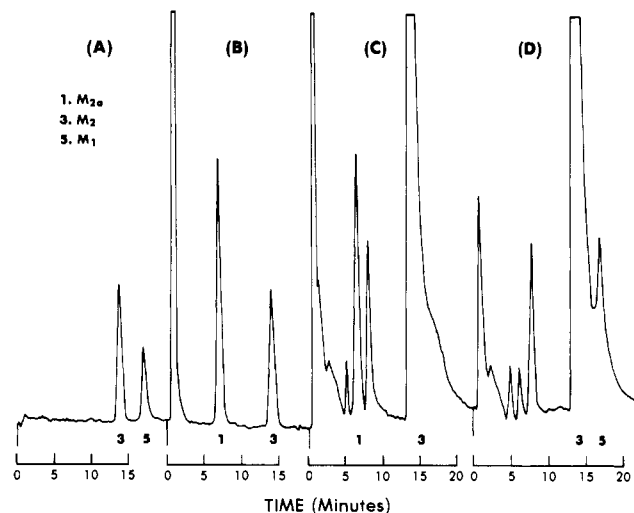


Figure 3. LC determination of 1 ppb aflatoxin M₁ in milk: (A) standards untreated with TFA, (B) standards with 15-min TFA reaction, (C) milk sample with TFA, (D) milk sample, no TFA.

controlled feeding study, there appears to be definite correlation between B₁ feeding levels and M₁ residue levels.

A limited recovery study was done using one sample each of sour cream, buttermilk, and milk. At the level of 1 ppb added M₁, the recoveries were 110, 125, and 93%.

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